

# **Dy-Mark Turfmaster White Dy-Mark**

Chemwatch: 4502-66 Version No: 13.1

Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements

### Chemwatch Hazard Alert Code: 4

Issue Date: 20/08/2021 Print Date: 27/04/2022 S.GHS.AUS.EN.E

### SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier	
Product name	Dy-Mark Turfmaster White
Chemical Name	Not Applicable
Synonyms	41125011
Proper shipping name	AEROSOLS
Chemical formula	Not Applicable
Other means of identification	Not Available

### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Aerosol paint.
	Application is by spray atomisation from a hand held aerosol pack

### Details of the supplier of the safety data sheet

Registered company name	y-Mark	
Address	89 Formation Street Wacol QLD 4076 Australia	
Telephone	1 7 3327 3004	
Fax	+61 7 3327 3009	
Website	http://www.dymark.com.au	
Email	info@dymark.com.au	

### Emergency telephone number

Association / Organisation	Dy-Mark
Emergency telephone numbers	+61 7 3327 3099
Other emergency telephone numbers	Not Available

# **SECTION 2 Hazards identification**

### Classification of the substance or mixture

### HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

## ChemWatch Hazard Ratings



Poisons Schedule	Not Applicable	
Classification [1]	Aerosols Category 1, Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 2A, Specific Target Organ Toxicity - Single Exposure (Narcotic Effects) Category 3	
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI	

## Label elements

### **Dy-Mark Turfmaster White**

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### Hazard pictogram(s)





Signal word
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word Danger

### Hazard statement(s)

AUH044	isk of explosion if heated under confinement.	
H222+H229	Extremely flammable aerosol. Pressurized container: may burst if heated.	
H315	Causes skin irritation.	
H319	Causes serious eye irritation.	
H336	May cause drowsiness or dizziness.	

### Precautionary statement(s) Prevention

P211	Do not spray on an open flame or other ignition source.  Do not pierce or burn, even after use.	
P271	Use only outdoors or in a well-ventilated area.	
P261	Avoid breathing mist/vapours/spray.	
P280	Wear protective gloves, protective clothing, eye protection and face protection.	
P264	Wash all exposed external body areas thoroughly after handling.	

### Precautionary statement(s) Response

P305+P351+P338	IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P312	Call a POISON CENTER/doctor/physician/first aider/if you feel unwell.	
P337+P313	If eye irritation persists: Get medical advice/attention.	
P302+P352	ON SKIN: Wash with plenty of water.	
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.	
P332+P313	If skin irritation occurs: Get medical advice/attention.	
P362+P364	Take off contaminated clothing and wash it before reuse.	

### Precautionary statement(s) Storage

	· · · · · · · · · · · · · · · · · · ·	
P405	Store locked up.	
P410+P412	Protect from sunlight. Do not expose to temperatures exceeding 50 °C/122 °F.	
P403+P233 Store in a well-ventilated place. Keep container tightly closed.		

### Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

### **SECTION 3 Composition / information on ingredients**

### Substances

See section below for composition of Mixtures

### Mixtures

CAS No	%[weight]	Name
64-17-5	10-30	ethanol
107-98-2	10-30	propylene glycol monomethyl ether - alpha isomer
Not Available	balance	Ingredients determined not to be hazardous
115-10-6	30-60	dimethyl ether
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available	

### **SECTION 4 First aid measures**

### Description of first aid measures

**Eye Contact** 

If aerosols come in contact with the eyes:

Immediately hold the eyelids apart and flush the eye continuously for at least 15 minutes with fresh running water.

# • Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.

- ► Transport to hospital or doctor without delay.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

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Skin Contact	If solids or aerosol mists are deposited upon the skin:  Flush skin and hair with running water (and soap if available).  Remove any adhering solids with industrial skin cleansing cream.  DO NOT use solvents.  Seek medical attention in the event of irritation.
Inhalation	If aerosols, fumes or combustion products are inhaled:
Ingestion	Not considered a normal route of entry.

### Indication of any immediate medical attention and special treatment needed

I reat symptomatically.
for lower alkyl ethers:
BASIC TREATMENT

Establish a patent airway with suction where necessary.

- Watch for signs of respiratory insufficiency and assist ventilation as necessary
- Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- A low-stimulus environment must be maintained.
- Monitor and treat, where necessary, for shock.
- Anticipate and treat, where necessary, for seizures.
- ▶ DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.

# ADVANCED TREATMENT

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- F Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Figure 10 Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- ▶ Hypotension without signs of hypovolaemia may require vasopressors.
- Treat seizures with diazepam
- Proparacaine hydrochloride should be used to assist eye irrigation.

### EMERGENCY DEPARTMENT

Laboratory analysis of complete blood count, serum electrolytes, BUN, creatinine, glucose, urinalysis, baseline for serum aminotransferases (ALT and AST), calcium, phosphorus and magnesium, may assist in establishing a treatment regime. Other useful analyses include anion and osmolar gaps, arterial blood gases (ABGs), chest radiographs and electrocardiograph.

- Ethers may produce anion gap acidosis. Hyperventilation and bicarbonate therapy might be indicated
- Haemodialysis might be considered in patients with impaired renal function.
- Consult a toxicologist as necessary.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

For acute or short term repeated exposures to ethanol:

- Acute ingestion in non-tolerant patients usually responds to supportive care with special attention to prevention of aspiration, replacement of fluid and correction of nutritional deficiencies (magnesium, thiamine pyridoxine, Vitamins C and K).
- Give 50% dextrose (50-100 ml) IV to obtunded patients following blood draw for glucose determination.
- Comatose patients should be treated with initial attention to airway, breathing, circulation and drugs of immediate importance (glucose, thiamine).
- Decontamination is probably unnecessary more than 1 hour after a single observed ingestion. Cathartics and charcoal may be given but are probably not effective in single ingestions.
- Fructose administration is contra-indicated due to side effects.

### **SECTION 5 Firefighting measures**

### **Extinguishing media**

SMALL FIRE:

► Water spray, dry chemical or CO2

LARGE FIRE:

Water spray or fog.

### Special hazards arising from the substrate or mixture

Fire Incompatibility

Fire Fighting

▶ Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

### Advice for firefighters

- Alert Fire Brigade and tell them location and nature of hazard.
- May be violently or explosively reactive.
- Wear breathing apparatus plus protective gloves.
- ▶ Prevent, by any means available, spillage from entering drains or water course.
- If safe, switch off electrical equipment until vapour fire hazard removed.
- Use water delivered as a fine spray to control fire and cool adjacent area.
- ▶ DO NOT approach containers suspected to be hot.
- Cool fire exposed containers with water spray from a protected location.
- If safe to do so, remove containers from path of fire.
- Equipment should be thoroughly decontaminated after use.

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### **Dy-Mark Turfmaster White**

Liquid and vapour are highly flammable. Severe fire hazard when exposed to heat or flame.
Vapour forms an explosive mixture with air.  $\mbox{\Large \ \ }$  Severe explosion hazard, in the form of vapour, when exposed to flame or spark. ▶ Vapour may travel a considerable distance to source of ignition ▶ Heating may cause expansion or decomposition with violent container rupture. ▶ Aerosol cans may explode on exposure to naked flames. Fire/Explosion Hazard ▶ Rupturing containers may rocket and scatter burning materials. Hazards may not be restricted to pressure effects. May emit acrid, poisonous or corrosive fumes. ▶ On combustion, may emit toxic fumes of carbon monoxide (CO). Combustion products include: carbon dioxide (CO2) other pyrolysis products typical of burning organic material.

Contains low boiling substance: Closed containers may rupture due to pressure buildup under fire conditions. **HAZCHEM** Not Applicable

### **SECTION 6 Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

See section 8

### **Environmental precautions**

See section 12

### Methods and material for containment and cleaning up

Methods and material for conta	ainment and cleaning up
Minor Spills	<ul> <li>Clean up all spills immediately.</li> <li>Avoid breathing vapours and contact with skin and eyes.</li> <li>Wear protective clothing, impervious gloves and safety glasses.</li> <li>Shut off all possible sources of ignition and increase ventilation.</li> <li>Wipe up.</li> <li>If safe, damaged cans should be placed in a container outdoors, away from all ignition sources, until pressure has dissipated.</li> <li>Undamaged cans should be gathered and stowed safely.</li> </ul>
Major Spills	<ul> <li>After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.</li> <li>Remove leaking cylinders to a safe place if possible.</li> <li>Release pressure under safe, controlled conditions by opening the valve.</li> <li>DO NOT exert excessive pressure on valve; DO NOT attempt to operate damaged valve.</li> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>May be violently or explosively reactive.</li> <li>Wear breathing apparatus plus protective gloves.</li> <li>Prevent, by any means available, spillage from entering drains or water courses</li> <li>No smoking, naked lights or ignition sources.</li> <li>Increase ventilation.</li> <li>Stop leak if safe to do so.</li> <li>Water spray or fog may be used to disperse / absorb vapour.</li> <li>Absorb or cover spill with sand, earth, inert materials or vermiculite.</li> <li>If safe, damaged cans should be placed in a container outdoors, away from ignition sources, until pressure has dissipated.</li> <li>Undamaged cans should be gathered and stowed safely.</li> <li>Collect residues and seal in labelled drums for disposal.</li> </ul>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and st	orage
Precautions for safe handling	
Safe handling	<ul> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> <li>DO NOT enter confined spaces until atmosphere has been checked.</li> <li>Avoid smoking, naked lights or ignition sources.</li> <li>Avoid contact with incompatible materials.</li> <li>When handling, DO NOT eat, drink or smoke.</li> <li>DO NOT incinerate or puncture aerosol cans.</li> <li>DO NOT spray directly on humans, exposed food or food utensils.</li> <li>Avoid physical damage to containers.</li> <li>Always wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately.</li> <li>Use good occupational work practice.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.</li> </ul>
Other information	<ul> <li>Keep dry to avoid corrosion of cans. Corrosion may result in container perforation and internal pressure may eject contents of can</li> <li>Store in original containers in approved flammable liquid storage area.</li> <li>DO NOT store in pits, depressions, basements or areas where vapours may be trapped.</li> <li>No smoking, naked lights, heat or ignition sources.</li> <li>Keep containers securely sealed. Contents under pressure.</li> <li>Store away from incompatible materials.</li> <li>Store in a cool, dry, well ventilated area.</li> <li>Avoid storage at temperatures higher than 40 deg C.</li> <li>Store in an upright position.</li> <li>Protect containers against physical damage.</li> </ul>

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- Check regularly for spills and leaks.
- ▶ Observe manufacturer's storage and handling recommendations contained within this SDS.

### Conditions for safe storage, including any incompatibilities

### Suitable container

- Aerosol dispenser.
- ► Check that containers are clearly labelled

### Storage incompatibility

Avoid reaction with oxidising agentsAvoid strong bases.













- X Must not be stored together
- 0 May be stored together with specific preventions
- May be stored together

Note: Depending on other risk factors, compatibility assessment based on the table above may not be relevant to storage situations, particularly where large volumes of dangerous goods are stored and handled. Reference should be made to the Safety Data Sheets for each substance or article and risks assessed accordingly.

### **SECTION 8 Exposure controls / personal protection**

### **Control parameters**

### Occupational Exposure Limits (OEL)

### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	ethanol	Ethyl alcohol	1000 ppm / 1880 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	propylene glycol monomethyl ether - alpha isomer	Propylene glycol monomethyl ether	100 ppm / 369 mg/m3	553 mg/m3 / 150 ppm	Not Available	Not Available
Australia Exposure Standards	dimethyl ether	Dimethyl ether	400 ppm / 760 mg/m3	950 mg/m3 / 500 ppm	Not Available	Not Available

### **Emergency Limits**

Ingredient	TEEL-1	TEEL-2	TEEL-3
ethanol	Not Available	Not Available	15000* ppm
propylene glycol monomethyl ether - alpha isomer	100 ppm	160 ppm	660 ppm
dimethyl ether	3,000 ppm	3800* ppm	7200* ppm

Ingredient	Original IDLH	Revised IDLH
ethanol	3,300 ppm	Not Available
propylene glycol monomethyl ether - alpha isomer	Not Available	Not Available
dimethyl ether	Not Available	Not Available

### **Exposure controls**

Appropriate engineering

controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

General exhaust is adequate under normal conditions. If risk of overexposure exists, wear SAA approved respirator. Correct fit is essential to obtain adequate protection.

Provide adequate ventilation in warehouse or closed storage areas.

Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

# Type of Contaminant: aerosols, (released at low velocity into zone of active generation) direct spray, spray painting in shallow booths, gas discharge (active generation into zone of rapid air motion) 1-2.5 m/s (200-500 f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

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Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min.) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

### Personal protection











# Eye and face protection

► Safety glasses with side shields

Chemical goggles

Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

### Skin protection

See Hand protection below

- ► Neoprene gloves
- ► No special equipment needed when handling small quantities.

### ▶ OTHERWISE:

### Hands/feet protection

- ► For potentially moderate exposures:
- Wear general protective gloves, eg. light weight rubber gloves.
- ► For potentially heavy exposures:
- ▶ Wear chemical protective gloves, eg. PVC. and safety footwear

### **Body protection**

See Other protection below

No special equipment needed when handling small quantities.

### OTHERWISE:

- Overalls.
- Skin cleansing cream.
- Evewash unit.
- Other protection Eyewash u
  - Do not spray on hot surfaces.
  - The clothing worn by process operators insulated from earth may develop static charges far higher (up to 100 times) than the minimum ignition energies for various flammable gas-air mixtures. This holds true for a wide range of clothing materials including cotton.
  - Avoid dangerous levels of charge by ensuring a low resistivity of the surface material worn outermost.

BRETHERICK: Handbook of Reactive Chemical Hazards.

### Recommended material(s)

# GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the: "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

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Material	СРІ
BUTYL	A
NEOPRENE	A
NATURAL RUBBER	С
NATURAL+NEOPRENE	С
NITRILE	С
NITRILE+PVC	С
PE/EVAL/PE	С
PVC	С

- \* CPI Chemwatch Performance Index
- A: Best Selection
- B: Satisfactory; may degrade after 4 hours continuous immersion
- C: Poor to Dangerous Choice for other than short term immersion

**NOTE**: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

### Respiratory protection

Type AX Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	AX-AUS / Class 1	-	AX-PAPR-AUS / Class 1
up to 50 x ES	Air-line*	-	-
up to 100 x ES	-	AX-3	-
100+ x ES	-	Air-line**	-

- \* Continuous-flow; \*\* Continuous-flow or positive pressure demand A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)
  - Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
  - The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
  - Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used
- Positive pressure, full face, air-supplied breathing apparatus should be used for work in enclosed spaces if a leak is suspected or the primary containment is to be opened (e.g. for a cylinder change)
- Air-supplied breathing apparatus is required where release of gas from primary containment is either suspected or demonstrated.

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### **SECTION 9 Physical and chemical properties**

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Appearance	Coloured liquid with a sweet solvent odour; partly miscible with water. Supplied as an aerosol pack. Contents under <b>PRESSURE</b> . Contains highly flammable hydrocarbon propellant.		
Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
nitial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	-41 (propellant)	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	HIGHLY FLAMMABLE.	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Partly miscible	pH as a solution (Not Available%)	Not Available

### **SECTION 10 Stability and reactivity**

Vapour density (Air = 1) Not Available

Reactivity	See section 7
Chemical stability	Elevated temperatures.     Presence of open flame.     Product is considered stable.     Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

### **SECTION 11 Toxicological information**

### Information on toxicological effects

Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.

There is some evidence to suggest that the material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.

VOC g/L Not Available

Inhalation of toxic gases may cause:

- ▶ Central Nervous System effects including depression, headache, confusion, dizziness, stupor, coma and seizures;
- respiratory: acute lung swellings, shortness of breath, wheezing, rapid breathing, other symptoms and respiratory arrest;
- ▶ heart: collapse, irregular heartbeats and cardiac arrest;
- gastrointestinal: irritation, ulcers, nausea and vomiting (may be bloody), and abdominal pain.

### Inhaled

Following inhalation, ethers cause lethargy and stupor. Inhaling lower alkyl ethers results in headache, dizziness, weakness, blurred vision, seizures and possible coma.

PGME has an offensive odour, and may cause drowsiness and unconsciousness if higher concentrations are inhaled, and severe reactions involving the eyes, nose and throat.

Material is highly volatile and may quickly form a concentrated atmosphere in confined or unventilated areas. The vapour may displace and replace air in breathing zone, acting as a simple asphyxiant. This may happen with little warning of overexposure.

Inhalation of high concentrations of gas/vapour causes lung irritation with coughing and nausea, central nervous depression with headache and dizziness, slowing of reflexes, fatigue and inco-ordination.

If exposure to highly concentrated solvent atmosphere is prolonged this may lead to narcosis, unconsciousness, even coma and possible death. WARNING: Intentional misuse by concentrating/inhaling contents may be lethal.

Accidental ingestion of the material may be damaging to the health of the individual.

Not normally a hazard due to physical form of product.

### Considered an unlikely route of entry in commercial/industrial environments Ingestion

Propylene glycol monomethyl ether has low hazard if taken orally. Ingestion of large amounts may cause headache, nausea, vomiting, diarrhoea, light-headedness, drowsiness, inco-ordination, CNS depression, kidney and liver injury in rats, unconsciousness, stoppage of breathing and possible death from anaesthesia.

# **Skin Contact**

The material may accentuate any pre-existing dermatitis condition

Repeated exposure may cause skin cracking, flaking or drying following normal handling and use.

Skin contact with the material may damage the health of the individual; systemic effects may result following absorption.

Spray mist may produce discomfort

Alkyl ethers may defat and dehydrate the skin producing dermatoses. Absorption may produce headache, dizziness, and central nervous system

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deceasion.    Hamila almanus of PCME may be absorbed investy the skin following extensive prolanged contact; this may result in discretizes.    Lower Concession and Content of the American Content of the Content of						
refleces and learn.  The introduction for a supposed region inflation in some persons and produce eye damage 24 fours or more after instillation. Geven The supposed of the control of the		Harmful amounts of PGME may be absorbed through the skin following extensive prolonged contact; this may result in drowsiness, unconsciousness and depression.  Open cuts, abraded or irritated skin should not be exposed to this material  There is some evidence to suggest that the material may cause moderate inflammation of the skin either following direct contact or after a delay				
Main rouse of exposure to the gas in the exciptions is by inhalation.  Chronic copyours to ally fairs may roused in roles of appetite, excessive thirst, latigue, and weight loss.  When taken represently, POSIC may cause durings to liver and string, disrections and even unconsciousness and death. There is no evidence construction in dogs. Ariental resting also string the programment in the same and resting, to depen destination in dogs. Ariental resting also structured in the production of the	Еуе	Not considered to be a risk because of the extreme volatility of the gas. Eye contact with alkyl ethers (vapour or liquid) may produce irritation, redness and tears.  There is evidence that material may produce eye irritation in some persons and produce eye damage 24 hours or more after instillation. Severe inflammation may be expected with pain.  Direct contact of the eye with ethanol (alcohol) may cause an immediate stinging and burning sensation, with reflex closure of the lid, and a temporary, tearing injury to the cornea together with redness of the conjunctiva. Discomfort may last 2 days but usually the injury heals without				
## Dy-Mark Turlmaster White  ## PropyLene glycol monomethyl either - alpha isomer  ## TOXICITY    Institution(Plan   L.550 - 6000 ppm4h <sup>2</sup>     Corl (Rat) LD50: 7000 mg/kg <sup>2</sup>     Eye (rabbil) 1000 mg/24th moderate   Eye (rabbil) 1000 mg/24th moderate   Eye adverse effect observed (instating) <sup>[1]</sup>   Sixin (rabbil) 200 mg/24th moderate   Sixin rabbil 200 mg/24th moderate   Corl (Rat) LD50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Corl (Rat) LD50: -3000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Corl (Rat) LD50: -3000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -20	Chronic	Main route of exposure to the gas in the workplace is by inhalation. Chronic exposure to alkyl ethers may result in loss of appetite, excessive thirst, fatigue, and weight loss. When taken repeatedly, PGME may cause damage to liver and kidney, drowsiness and even unconsciousness and death. There is no evidence of damage to the sex organs. However, it has led to multiple pregnancies in rats and rabbits, but sperm destruction in dogs. Animal testing also shows high doses can delay bone development. Some glycol esters and their ethers cause wasting of the testicles, reproductive changes, infertility and changes to kidney function. Shorter chain				
## Dy-Mark Turlmaster White  ## PropyLene glycol monomethyl either - alpha isomer  ## TOXICITY    Institution(Plan   L.550 - 6000 ppm4h <sup>2</sup>     Corl (Rat) LD50: 7000 mg/kg <sup>2</sup>     Eye (rabbil) 1000 mg/24th moderate   Eye (rabbil) 1000 mg/24th moderate   Eye adverse effect observed (instating) <sup>[1]</sup>   Sixin (rabbil) 200 mg/24th moderate   Sixin rabbil 200 mg/24th moderate   Corl (Rat) LD50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Corl (Rat) LD50: -3000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Corl (Rat) LD50: -3000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -2000 mg/kg <sup>2</sup>     Inhalation(Plan) LC50: -20		TOXICITY	IRRITATION			
ethanol  ethanol  ethanol  Dermal (rabbit) LDSc: 7100 mg/kg <sup>[1]</sup> Eye (rabbit): 500 mg SEVERE    Inhalation(Rat) LDSc: 64000 ppm/hil <sup>2</sup>   Eye: (rabbit): 100 mg/s24he-moderate   Skin (rabbit): 200 mg/s24he-moderate   Skin (rabbit): 400 mg (penp-lmid)   Skin (rabbit): 500 mg 24 h mild   Oral (Rat) LDSc: 2000 mg/kg <sup>[1]</sup> Eye (rabbit): 500 mg 24 h mild   Oral (Rat) LDSc: 3739 mg/kg <sup>[1]</sup> Eye (rabbit): 500 mg 24 h mild   TOXICITY   Inhalation(Rat) LCSc: 20000 ppm/hil <sup>1</sup> Not Available    Legend:   1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise aprecified date extracted from #TESCS - Register of Toxic Effect of chemical Substances   PROPYLENG CLYCOL   NONMETHY, ETHER. ALPHA ISOMER   Not Applicated to the miles of the substances of not proportion by the substances of not proportion of vision in the substance of not proportion gives a but not in rabbits at this concentration; maternal toxicity was noted in both species. Science of Proportions (PRES)   Special proportions gived in the substance of the must proportion gived between seein in rabb ton foil in abbits at this concentration; maternal toxicity was noted in both species. Science in the development place of the proportions and science of the bover molecular weight homologue in the development place of the proportions and ethosysactic adds.   PROPYLENG CLYCOL MONMETHY, ETHER. ALPHA ISOMER   Proportions of the development place of the proportions and ethosysactic adds.   PROPYLE	Dy-Mark Turfmaster White					
ethanol  ethanol  ethanol  Dermal (rabbit) LDSc: 7100 mg/kg <sup>[1]</sup> Eye (rabbit): 500 mg SEVERE    Inhalation(Rat) LDSc: 64000 ppm/hil <sup>2</sup>   Eye: (rabbit): 100 mg/s24he-moderate   Skin (rabbit): 200 mg/s24he-moderate   Skin (rabbit): 400 mg (penp-lmid)   Skin (rabbit): 500 mg 24 h mild   Oral (Rat) LDSc: 2000 mg/kg <sup>[1]</sup> Eye (rabbit): 500 mg 24 h mild   Oral (Rat) LDSc: 3739 mg/kg <sup>[1]</sup> Eye (rabbit): 500 mg 24 h mild   TOXICITY   Inhalation(Rat) LCSc: 20000 ppm/hil <sup>1</sup> Not Available    Legend:   1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise aprecified date extracted from #TESCS - Register of Toxic Effect of chemical Substances   PROPYLENG CLYCOL   NONMETHY, ETHER. ALPHA ISOMER   Not Applicated to the miles of the substances of not proportion by the substances of not proportion of vision in the substance of not proportion gives a but not in rabbits at this concentration; maternal toxicity was noted in both species. Science of Proportions (PRES)   Special proportions gived in the substance of the must proportion gived between seein in rabb ton foil in abbits at this concentration; maternal toxicity was noted in both species. Science in the development place of the proportions and science of the bover molecular weight homologue in the development place of the proportions and ethosysactic adds.   PROPYLENG CLYCOL MONMETHY, ETHER. ALPHA ISOMER   Proportions of the development place of the proportions and ethosysactic adds.   PROPYLE			'			
tribulation(Rat) LC50; e4000 pmr4  <sup>[5]</sup> Eye (rabbit):100mg/24hr-moderate    Propylene glycol monomethyl ether - alpha isomer			-			
ethanol  Cral (Rat) LD50; 7050 mg/kg <sup>[2]</sup> Eye: adverse effect observed (irritating) <sup>[1]</sup> Skin (rabbit); 400 mg (open)-mild  Skin: no adverse effect observed (not irritating) <sup>[1]</sup> TOXICITY  IRRITATION  dermal (rat) LD50; >2000 mg/kg <sup>[1]</sup> Eye (rabbit) :300 mg mild  inhalation(Rat) LC50; >6 mg/kh <sup>[2]</sup> Cral (Rat) LD50; 3739 mg/kg <sup>[1]</sup> Eye (rabbit) :500 mg/24 h - mild  TOXICITY  dimethyl ether  Alepand:  I Value obtained from Europe ECHA Registered Substances - Acute toxicity 2* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances  ETHANOL  PROPYLENE GLYCOL  MONOMETHYL ETHER- ALPHA ISOMER  Dy-Mark Turfmaster White & Propose of the Composition of the Series. In the ethylene series, In a common toxicities associated with the lower molecular weight normal page of them common toxicities associated with the lower molecular weight normal page of them of the lower molecular weight normal page of them of the lower molecular weight normal page of them of the lower molecular weight normal page of them of the lower molecular weight normal page of them of the lower molecular weight normal page of them of the lower molecular weight normal page of them of the lower molecular weight homologues of the enthylene series. In the ethylene series, under a natively of proprise glycol eithers in action of the molecular page of the PGEs in broad on the lower molecular weight homologues of the ethylene series, such as adverse effects on the proproductive organs, the developing membry and foetas, blood or thymus gland, are not seen with the PGEs which is the some reproductive organs can be a natively proprise glycol of thems, longer having the more page of the page		Dermal (rabbit) LD50: 17100 mg/kg <sup>[1]</sup>				
Skin (rabbit):20 mg/24hr-moderate   Skin (rabbit):20 mg/24hr-moderate   Skin (rabbit):20 mg/24hr-moderate   Skin (rabbit):400 mg (open-mid   Skin: no adverse effect observed (not irritating) <sup>[1]</sup>		Inhalation(Rat) LC50; 64000 ppm4h <sup>[2]</sup>	Eye (rabbit):100	mg/24hr-moderate		
Skin (rabbit)-400 mg (opon)-mild   Skin: no adverse effect observed (not irritating) <sup>[1]</sup>	ethanol	Oral (Rat) LD50; 7060 mg/kg <sup>[2]</sup>	-			
TOXICITY   IRRITATION   demail (rat) LD50: >2000 mg/kg <sup>[1]</sup>   Eye (rabbit) 230 mg mild   lnhalation(Rat) LD50: >2000 mg/kg <sup>[1]</sup>   Eye (rabbit) 500 mg/24 h mild   mlalation(Rat) LD50: >5 mg/kg <sup>[1]</sup>   Eye (rabbit) 500 mg/24 h mild   mlalation(Rat) LD50: >5 mg/kg <sup>[1]</sup>   Eye (rabbit) 500 mg/24 h mild   Skin (rabbit) 500 mg open - mild   Skin (rabbit) 500 mg						
propylene glycol monomethyle either - alpha isomer  ### CTANICITY    dermal (rat) LD50: >2000 mg/kg <sup>11</sup>   Eye (rabbit) 230 mg mild     Inhalation(Rat) LC50: >6 mg/kh <sup>20</sup>   Eye (rabbit) 500 mg 24 h mild     Inhalation(Rat) LC50: >6 mg/kh <sup>20</sup>   Eye (rabbit) 500 mg 24 h mild     TOXICITY   IRRITATION     Inhalation(Rat) LC50: >20000 ppm/kg <sup>11</sup>   Eye (rabbit) 500 mg open - mild     TOXICITY   IRRITATION     Inhalation(Rat) LC50: >20000 ppm/kg <sup>11</sup>   Eye (rabbit) 500 mg open - mild     TOXICITY   Inhalation(Rat) LC50: >20000 ppm/kg <sup>11</sup>   Not Available     Legend:   1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2: * Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances						
dermal (rat) LD50: >2000 mg/kg <sup>[1]</sup> Eye (rabbit) 230 mg mild   Inhalation(Rat) LC50: >6 mg/l4h <sup>[2]</sup> Eye (rabbit) 500 mg/24 h mild   Oral (Rat) LD50: 3739 mg/kg <sup>[1]</sup> Eye (rabbit) 500 mg/24 h mild   Oral (Rat) LD50: 3739 mg/kg <sup>[1]</sup> Eye (rabbit) 500 mg/24 h mild   TOXICITY   Inhalation(Rat) LC50: >20000 ppm4h <sup>[1]</sup> Not Available   Legend:   I. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Registered Substances and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.  PROPYLENE GLYCOL MONOMETHYLE THER - ALPHA ISOMER   NOTE: For PGE - mixed isomers: Exposure of pregnant rats and rabbits to the substance did not give rise to teratogenic effects at concentrations up to 3000 ppm. Foetotoxic effects were seen in rats but not in rabbits at this concentration; maternal toxicity was noted in both species.  PROPYLENE GLYCOL MONOMETHYLE THER - ALPHA ISOMER   Dy-Mark Turfmaster White & PROPYLENE GLYCOL MONOMETHYL ETHER - ALPHA ISOMER   PROPYLENE GLYCOL MONOMETHYL ETHER - ALPHA ISOMER   Propylene glycol of the breaking great of propylene glycol methyl ether (PMB): dipropylene glycol reposition of the terminal propylene glycol ethers has shown that propylene glycol because weight propylene glycol ethers in the ethylene series and associated with the lower molecular weight homologues in the ethylene series are due specifically to the formation of methodysaerola and ethoxyaerolic acid. Unique chain homologues in the ethylene series are due specifically to the formation of methodysaerola and ethoxyaerolic acids. Unique chain homologues in the ethylene series are not associ			Skin: no adverse	e effect observed (not irritating)[1]		
Inhalation(Rat) LC50; >6 mg/l4hl <sup>2</sup>   Eye (rabbit) 500 mg/24 h mild		TOXICITY	IRRITATION			
ether - alpha isomer  Oral (Rat) LD50; 3739 mg/kg <sup>11</sup>		dermal (rat) LD50: >2000 mg/kg <sup>[1]</sup>	Eye (rabbit) 230	mg mild		
Cral (Rat) LD50; 3739 mg/kg <sup>[1]</sup>   Eye (rabbit): 100 mg SEVERE						
TOXICITY   IRRITATION   Not Available	ether - alpha isomer					
Inhalation(Rat) LCS0; >20000 ppm4h <sup>[1]</sup> Not Available  Legend: 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances  The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.  PROPYLENE GLYCOL MONOMETHYL ETHER-ALPHA ISOMER  For propylene glycol ethers (PGEs): Typical propylene glycol ethers include pryolene glycol nebtyl ether (PB); dipropylene glycol nebtyl ether acetate (DPMA) and tripropylene glycol ethers include pryolene glycol ethers (PGEs): Typical propylene glycol ethers include pryolene glycol ethers (PGP); and the ethylene series, are to associated with reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol ethers in the ethylene series. Breatbolisms of the terminal hydroxy group produces and ethysaecetic acids. Longer chain homologues in the ethylene series are to associated with reproductive toxicity, but can cause haemolysis in sensitive species, also manufacture of PGEs) is a secondary alcohol incapable of forming an alkoxyaporiopionic acid. In contrast, beta-isomers are able to form the alloxyspropionic acids and these are linked to birth defects (and possibly, haemolytic effects). The alpha isomer comprises more than 95% of the isomeric mixture in the commercial product, and therefore PGEs show relatively little toxicity. One of the main metabolities of the propylene glycol ethers is propylene glycol toxicity and compropionic acid. In contrast, beta-isomers are able to form the alloxyspropionic acids and these are linked to birth defects (and possibly, haemolytic effects). The alpha isomer comprises more than 95% of the isomeric mixture in the commercial product, and therefore PGEs show relatively little toxicity. One of t		- Colar (Colar) 22 cos, co co mg mg				
Inhalation(Rat) LCS0; >20000 ppm4h <sup>[1]</sup> Not Available  Legend: 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances  The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.  PROPYLENE GLYCOL MONOMETHYL ETHER-ALPHA ISOMER  For propylene glycol ethers (PGEs): Typical propylene glycol ethers include pryolene glycol nebtyl ether (PB); dipropylene glycol nebtyl ether acetate (DPMA) and tripropylene glycol ethers include pryolene glycol ethers (PGEs): Typical propylene glycol ethers include pryolene glycol ethers (PGP); and the ethylene series, are to associated with reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol ethers in the ethylene series. Breatbolisms of the terminal hydroxy group produces and ethysaecetic acids. Longer chain homologues in the ethylene series are to associated with reproductive toxicity, but can cause haemolysis in sensitive species, also manufacture of PGEs) is a secondary alcohol incapable of forming an alkoxyaporiopionic acid. In contrast, beta-isomers are able to form the alloxyspropionic acids and these are linked to birth defects (and possibly, haemolytic effects). The alpha isomer comprises more than 95% of the isomeric mixture in the commercial product, and therefore PGEs show relatively little toxicity. One of the main metabolities of the propylene glycol ethers is propylene glycol toxicity and compropionic acid. In contrast, beta-isomers are able to form the alloxyspropionic acids and these are linked to birth defects (and possibly, haemolytic effects). The alpha isomer comprises more than 95% of the isomeric mixture in the commercial product, and therefore PGEs show relatively little toxicity. One of t			·			
Legend:  1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances  The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.  NOTE: For PGE - mixed isomers: Exposure of pregnant rats and rabbits to the substance did not give rise to teratogenic effects at concentrations up to 3000 ppm. Foetotoxic effects were seen in rats but not in rabbits at this concentration; maternal toxicity was noted in both species.  For propylene glycol ethers (PGEs): Typical propylene glycol ethers (PGEs): Typical propylene glycol ethers include propylene glycol n-butyl ether (PnB); dipropylene glycol n-butyl ether (DPnB); dipropylene glycol n-butyl ether acetate (DPMA) and tripropylene glycol ethers has shown that propylene glycol-based ethers are less toxic than some ethers of the ethylene series. The common toxicities associated with the lower molecular weight homologues of the ethylene series, such as adverse effects on the reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol ethers. In the ethylene series, metabolism of the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol ethers. In the ethylene series, metabolism of the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol. Monomercial productive toxicity, but can cause haemolysis in sensitive species, also through formation of an alkoxyacetic acid. The predominant apha isomer of all the PGEs (which is thermodynamically revoured	dimethyl ether					
ETHANOL  The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.  NOTE: For PGE - mixed isomers: Exposure of pregnant rats and rabbits to the substance did not give rise to teratogenic effects at concentrations up to 3000 ppm. Foetotoxic effects were seen in rats but not in rabbits at this concentration; maternal toxicity was noted in both species.  For propylene glycol ethers (PGEs): Typical propylene glycol ethers include propylene glycol n-butyl ether (PnB); dipropylene glycol n-butyl ether (DPnB); dipropylene glycol methyl ether acetate (DPMA) and tripropylene glycol ethers has shown that propylene glycol-based ethers are less toxic than some ethers of the ethylene series. The common toxicities associated with the lower molecular weight homologues of the ethylene series, such as adverse effects on the office of the lower molecular weight homologues in the ethylene series are ense with the commercial-grade propylene glycol ethers. In the ethylene series, metabolism of the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive and developmental toxicities of the lower molecular weight homologues in the ethylene series are ense with the commercial-grade propylene glycol ethers. In the ethylene series, metabolism of the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive and developmental toxicities of the both monologues in the ethylene series are of alkoxyacetic acid. The reproductive toxicity, but can cause haemolysis in sensitive species, also formation of an alkoxyacetic acid. The predominant alpha isomer of all the PGEs (which is thermodynamically favoured during manufacture of PGEs) is a secondary alcohol incapable of forming an alkoxypropionic acid. In contrast, beta-isomers are able to form the alkoxypropionic acids and these are linked to birth defects (and possibly, haemolylic) effects). The alpha isomer comprises more than 95% o		Inhalation(Rat) LC50; >20000 ppm4h <sup>[1]</sup>	Not Available			
PROPYLENE GLYCOL MONOMETHYL ETHER- ALPHA ISOMER  Propylene glycol ethers (PGEs):  Typical propylene glycol ethers include propylene glycol n-butyl ether (PnB); dipropylene glycol methyl ether acetate (PDMA) and tripropylene glycol methyl ether (PnB); dipropylene glycol rebutyl ether acetate (PDMA) and tripropylene glycol methyl ether (PnB); dipropylene glycol n-butyl ether series, such as adverse effects on the reproductive organs, the developing embryo and foetus, blood or trymus gland, are not seen with the comercial-grade propylene for the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive and developmental toxicities of the lower molecular weight needs and through formation of an alkoxyacetic acid. The predominant alpha isomer of all the PGEs (which is thermodynamically favoured during manifecture of PGEs) is a secondary alcohol incapable of formal and loxyacetic productive organical setting, while the remaining members of this category caused little or no eye irritation. Animal testing showed that repeat dosing caused few adverse effects. Animal testing also shows that PGEs do not cause skin effects or reproductive toxicity. Commercially available PGEs have not been shown to cause birth the before the remaining members of this category caused little or no eye irritation. None caused skin sensitization.  Acute Toxicity  Skin Irritation/Corrosion  **Carcinogenicity**  **Carcinogenicity**  **Skin Irritation/Corrosion**  **NEOPTION Township and the substance did not give rise to teratogenic acid. The teratogenic rate of the substance did not give rise to teratogenic effects at concentrations up to 3000 ppm. Foetotoxic effects at concentration in anterior proplene glycol ethers are less toxic than some ease to the terminal hydroxyl group produces and alkoyacetic acid. The reproductive toxicity, but can cause haemolysis in sensitive species, also through formation of an alkoxyacetic acid. The predominant alpha isomer of all the PGEs (which is thermodynamically favoured during malkoxypropion	Legend:			ained from manufacturer's SDS. Unless otherwise		
MONOMETHYL ETHER- ALPHA ISOMER  Por propylene glycol ethers (PGEs): Typical propylene glycol ethers include propylene glycol n-butyl ether (PnB); dipropylene glycol n-butyl ether (DPnB); dipropylene glycol nethyl ether acetate (DPMA) and tripropylene glycol ethers include propylene glycol methyl ether (TPM).  Testing of a wide variety of propylene glycol ethers has shown that propylene glycol-based ethers are less toxic than some ethers of the ethylene series. The common toxicities associated with the lower molecular weight homologues of the ethylene series, such as adverse effects on the reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol ethers. In the ethylene series, metabolism of the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive and developmental toxicities of the lower molecular weight homologues in the ethylene series are due specifically to the formation of methoxyacetic acids. Unoger chain homologues in the ethylene series are not associated with reproductive toxicity, but can cause haemolysis in sensitive species, also through formation of an alkoxyacetic acid. The predominant alpha isomer of all the PGEs (which is thermodynamically favoured during manufacture of PGEs) is a secondary alcohol incapable of forming an alkoxypropionic acid. In contrast, beta-isomers are able to form the alkoxypropionic acids and these are linked to birth defects (and possibly, haemolytic effects). The alpha isomer comprises more than 95% of the isomeric mixture in the commercial product, and therefore PGEs show relatively little toxicity. One of the main metabolities of the propylene glycol ethers is propylene glycol, which is of low toxicity and completely metabolized in the body.  As a class, PGEs have low acute toxicity via swallowing, skin exposure and inhalation. PnB and TPM are moderately irritating to the eyes, in animal testing, while the remaining members of this category caused little or no eye irritation	ETHANOL		d or repeated exposure and may produ	ce on contact skin redness, swelling, the production of		
Typical propylene glycol ethers include propylene glycol n-butyl ether (PnB); dipropylene glycol n-butyl ether (DPnB); dipropylene glycol methyl ether acetate (DPMA) and tripropylene glycol ethers has shown that propylene glycol-based ethers are less toxic than some ethers of the ethylene series. The common toxicities associated with the lower molecular weight homologues of the ethylene series, such as adverse effects on the reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol ethers. In the ethylene series, metabolism of the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive and developmental toxicities of the lower molecular weight homologues in the ethylene series are due specifically to the formation of methoxyacetic acids. Longer chain homologues in the ethylene series are not associated with reproductive toxicity, but can cause haemolysis in sensitive species, also through formation of an alkoxyacetic acid. The predominant alpha isomer of all the PGEs (which is thermodynamically favoured during manufacture of PGEs) is a secondary alcohol incapable of forming an alkoxypropionic acid. In contrast, beta-isomers are able to form the alkoxypropionic acids and these are linked to birth defects (and possibly, haemolytic effects). The alpha isomer comprises more than 95% of the isomeric mixture in the commercial product, and therefore PGEs show relatively little toxicity. One of the main metabolites of the propylene glycol with significant productive toxicity of the series is propylene glycol, which is of low toxicity and completely metabolized in the body.  As a class, PGEs have low acute toxicity via swallowing, skin exposure and inhalation. PnB and TPM are moderately irritating to the eyes, in animal testing, while the remaining members of this category caused little or no eye irritation. None caused skin sensitization.  Animal testing showed that repeat dosing caused few adverse effects. Animal testing also	MONOMETHYL ETHER -	, , ,		· ·		
Skin Irritation/Corrosion	PROPYLENE GLYCOL MONOMETHYL ETHER -	Typical propylene glycol ethers include propylene glycol n-butyl ether (PnB); dipropylene glycol n-butyl ether (DPnB); dipropylene glycol methyl ether acetate (DPMA) and tripropylene glycol methyl ether (TPM).  Testing of a wide variety of propylene glycol ethers has shown that propylene glycol-based ethers are less toxic than some ethers of the ethylene series. The common toxicities associated with the lower molecular weight homologues of the ethylene series, such as adverse effects on the reproductive organs, the developing embryo and foetus, blood or thymus gland, are not seen with the commercial-grade propylene glycol ethers. In the ethylene series, metabolism of the terminal hydroxyl group produces and alkoxyacetic acid. The reproductive and developmental toxicities of the lower molecular weight homologues in the ethylene series are due specifically to the formation of methoxyacetic acid ethoxyacetic acids. Longer chain homologues in the ethylene series are not associated with reproductive toxicity, but can cause haemolysis in sensitive species, also through formation of an alkoxyacetic acid. The predominant alpha isomer of all the PGEs (which is thermodynamically favoured during manufacture of PGEs) is a secondary alcohol incapable of forming an alkoxypropionic acid. In contrast, beta-isomers are able to form the alkoxypropionic acids and these are linked to birth defects (and possibly, haemolytic effects). The alpha isomer comprises more than 95% of the isomeric mixture in the commercial product, and therefore PGEs show relatively little toxicity. One of the main metabolites of the propylene glycol ethers is propylene glycol, which is of low toxicity and completely metabolized in the body.  As a class, PGEs have low acute toxicity via swallowing, skin exposure and inhalation. PnB and TPM are moderately irritating to the eyes, in animal testing, while the remaining members of this category caused little or no eye irritation. None caused skin sensitization.  Animal testing showed that repeat dosing caused				
	Acute Toxicity	×	Carcinogenicity	×		
Serious Eye Damage/Irritation   ✓ STOT - Single Exposure  ✓	Skin Irritation/Corrosion	✓	Reproductivity	×		
	Serious Eye Damage/Irritation	<b>~</b>	STOT - Single Exposure	<b>~</b>		

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 Dy-Mark Turfmaster White
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# Respiratory or Skin sensitisation Mutagenicity X STOT - Repeated Exposure X Aspiration Hazard X

# **SECTION 12 Ecological information**

### Toxicity

	Endpoint	Test Duration (hr)	Species	Value	Source
Dy-Mark Turfmaster White	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50(ECx)	96h	Algae or other aquatic plants	<0.001mg/L	4
. 11 1	LC50	96h	Fish	>100mg/l	2
ethanol	EC50	72h	Algae or other aquatic plants	275mg/l	2
	EC50	48h	Crustacea	>79mg/L	4
	EC50	96h	Algae or other aquatic plants	<0.001mg/L	4
	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50(ECx)	168h	Algae or other aquatic plants	>1000mg/l	1
propylene glycol monomethyl	LC50	96h	Fish	>=1000mg/l	2
ether - alpha isomer	EC50	72h	Algae or other aquatic plants	>500mg/l	2
	EC50	48h	Crustacea	23300mg/l	1
	EC50	96h	Algae or other aquatic plants	>1000mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
	NOEC(ECx)	48h	Crustacea	>4000mg/l	1
dimethyl ether	LC50	96h	Fish	1783.04mg/l	2
	EC50	48h	Crustacea	>4400mg/L	2
	EC50	96h	Algae or other aquatic plants	154.917mg/l	2
Legend:	Ecotox databas		A Registered Substances - Ecotoxicological Informat quatic Hazard Assessment Data 6. NITE (Japan) - Bi		

### DO NOT discharge into sewer or waterways.

### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
ethanol	LOW (Half-life = 2.17 days)	LOW (Half-life = 5.08 days)
propylene glycol monomethyl ether - alpha isomer	LOW (Half-life = 56 days)	LOW (Half-life = 1.7 days)
dimethyl ether	LOW	LOW

# Bioaccumulative potential

Ingredient	Bioaccumulation
ethanol	LOW (LogKOW = -0.31)
propylene glycol monomethyl ether - alpha isomer	LOW (BCF = 2)
dimethyl ether	LOW (LogKOW = 0.1)

### Mobility in soil

Ingredient	Mobility
ethanol	HIGH (KOC = 1)
propylene glycol monomethyl ether - alpha isomer	HIGH (KOC = 1)
dimethyl ether	HIGH (KOC = 1.292)

### **SECTION 13 Disposal considerations**

### Waste treatment methods

Product / Packaging disposal

- ► DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.

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- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
   Consult State Land Waste Management Authority for disposal.
- $\mbox{\ensuremath{\,^{\blacktriangleright}}}$  Discharge contents of damaged aerosol cans at an approved site.
- Allow small quantities to evaporate.
- $\begin{tabular}{ll} \begin{tabular}{ll} \beg$
- ▶ Bury residues and emptied aerosol cans at an approved site.

# **SECTION 14 Transport information**

# **Labels Required**



Marine Pollutant
HAZCHEM

Not Applicable

### Land transport (ADG)

zana tranoport (ABG)			
UN number	950		
UN proper shipping name	AEROSOLS		
Transport hazard class(es)	Class 2.1 Subrisk Not Applicable		
Packing group	Not Applicable		
Environmental hazard	Not Applicable		
Special precautions for user	Special provisions   63 190 277 327 344 381		

### Air transport (ICAO-IATA / DGR)

UN number	1950			
UN proper shipping name	Aerosols, flammable	Aerosols, flammable		
Transport hazard class(es)	ICAO/IATA Class ICAO / IATA Subrisk ERG Code	2.1 Not Applicable		
Packing group	Not Applicable	Not Applicable		
Environmental hazard	Not Applicable			
Special precautions for user	Cargo Only Maximum Passenger and Cargo Passenger and Cargo Passenger and Cargo			

# Sea transport (IMDG-Code / GGVSee)

UN number	1950	1950		
UN proper shipping name	AEROSOLS			
Transport hazard class(es)		2.1Not Applicable		
Packing group	Not Applicable	Not Applicable		
Environmental hazard	Not Applicable	Not Applicable		
Special precautions for user	EMS Number Special provisions Limited Quantities			

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

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Product name	Group
ethanol	Not Available
propylene glycol monomethyl ether - alpha isomer	Not Available
dimethyl ether	Not Available

### Transport in bulk in accordance with the ICG Code

Product name	Ship Type
ethanol	Not Available
propylene glycol monomethyl ether - alpha isomer	Not Available
dimethyl ether	Not Available

### **SECTION 15 Regulatory information**

### Safety, health and environmental regulations / legislation specific for the substance or mixture

### ethanol is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australian Inventory of Industrial Chemicals (AIIC)

### propylene glycol monomethyl ether - alpha isomer is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australian Inventory of Industrial Chemicals (AIIC)

### dimethyl ether is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -Schedule 5

Australian Inventory of Industrial Chemicals (AIIC)

### **National Inventory Status**

National Inventory	Status
Australia - AIIC / Australia Non-Industrial Use	Yes
Canada - DSL	Yes
Canada - NDSL	No (ethanol; propylene glycol monomethyl ether - alpha isomer; dimethyl ether)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	Yes
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - FBEPH	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

### **SECTION 16 Other information**

Revision Date	20/08/2021
Initial Date	15/12/2001

### **SDS Version Summary**

Version	Date of Update	Sections Updated
12.1	01/11/2019	One-off system update. NOTE: This may or may not change the GHS classification
13.1	20/08/2021	Classification change due to full database hazard calculation/update.

# Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

### **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

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IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit $_{\circ}$ 

IDLH: Immediately Dangerous to Life or Health Concentrations

ES: Exposure Standard OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

AIIC: Australian Inventory of Industrial Chemicals

DSL: Domestic Substances List NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances
ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers

ENCS: Existing and New Chemical Substances Inventory

KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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TEL (+61 3) 9572 4700.